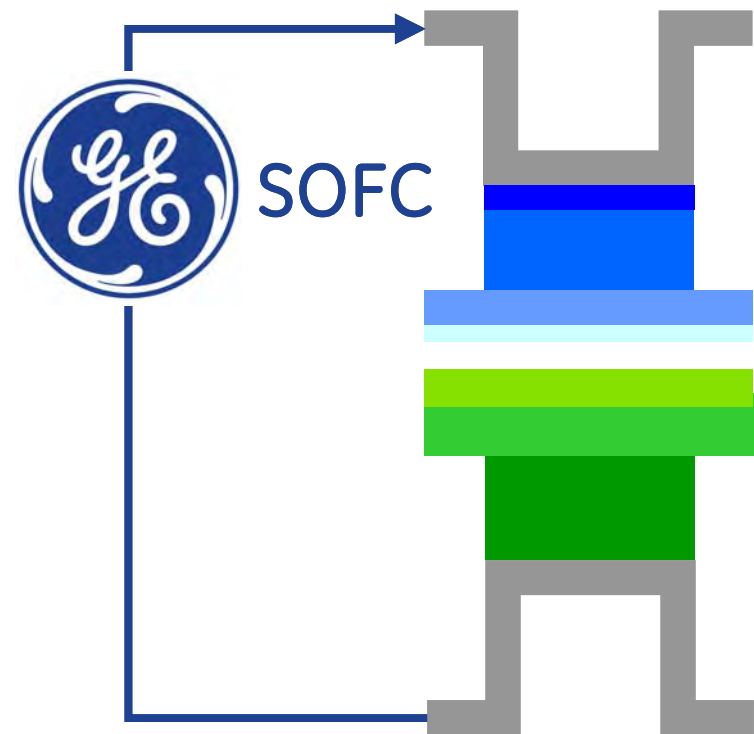


# SOFC Development at GE Global Research

Matt Alinger  
GE Global Research  
Niskayuna, NY

12<sup>th</sup> Annual SECA Workshop  
Pittsburgh, PA  
July 26-28, 2011

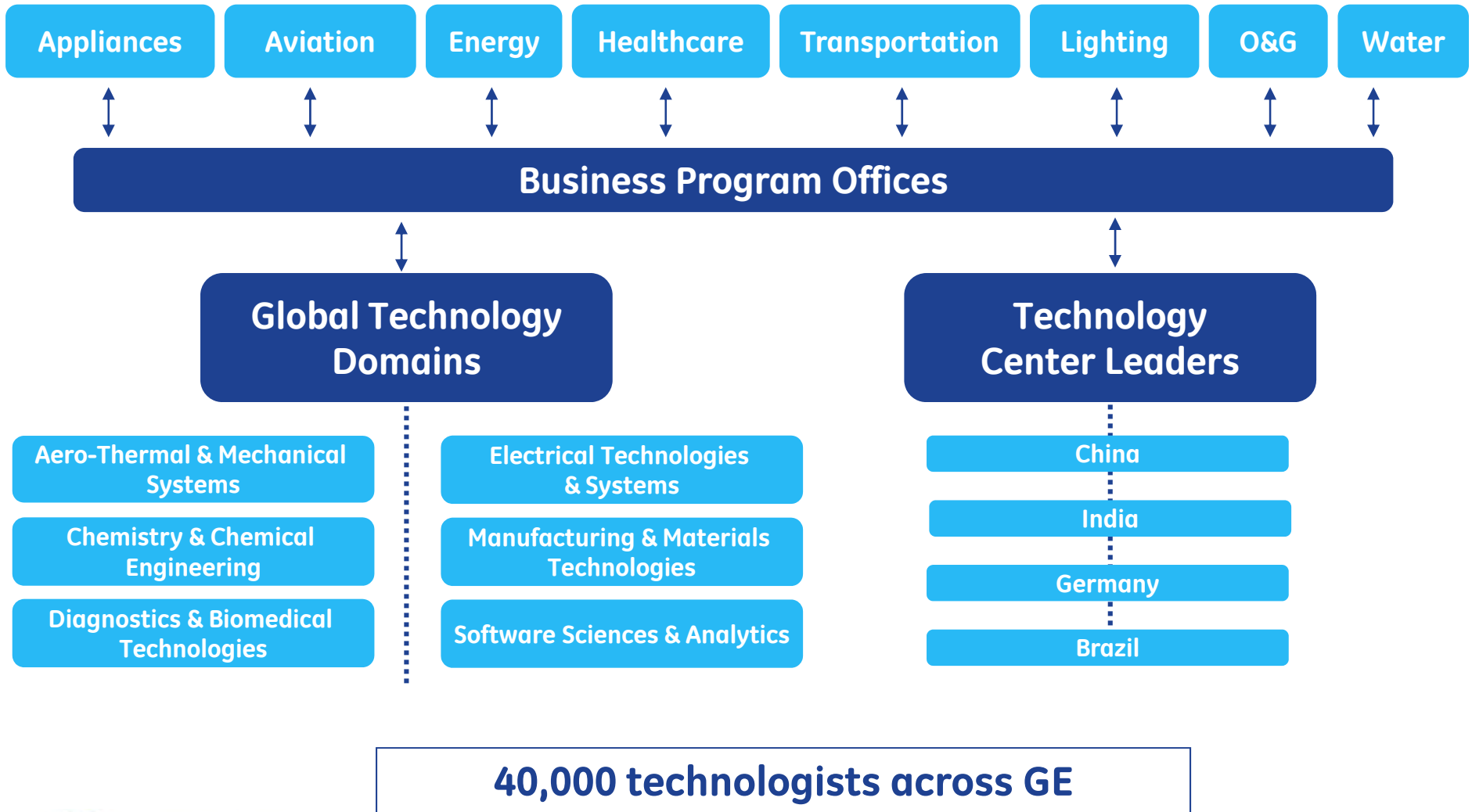


# GE today



Aligned for growth

# Global reach and connectivity



# Global Research annual funding

## GE business programs

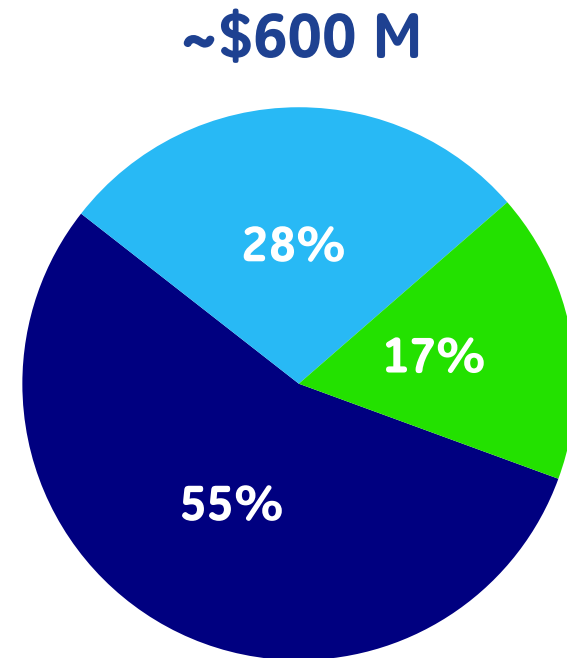
- Next generation product technology
- Short-term technical challenges

## GE corporate programs

- Advanced Technology programs
- New ideas
- High-risk/high reward

## External partnerships and gov't. funded

- Joint technology
- Specific customer focus

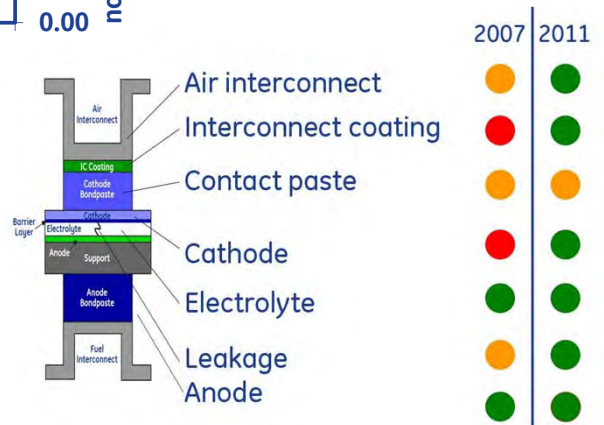
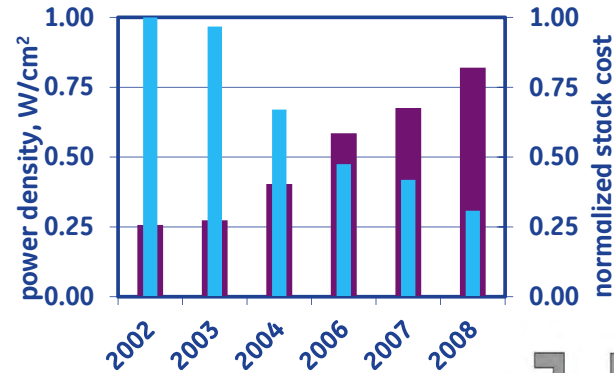


# Technology Challenges for SOFCs

- Materials set
  - High power density
  - Low degradation / stable
- Scale-up
  - Larger cells
  - Bigger stacks
- System design & integration
  - Improved design for reliability
  - Operability (start-up, shut down, transients, ...)

# Presentation Outline

- Materials Set
  - Cost
  - Performance
  - Degradation
  
- Scale-up
  - Manufacturing



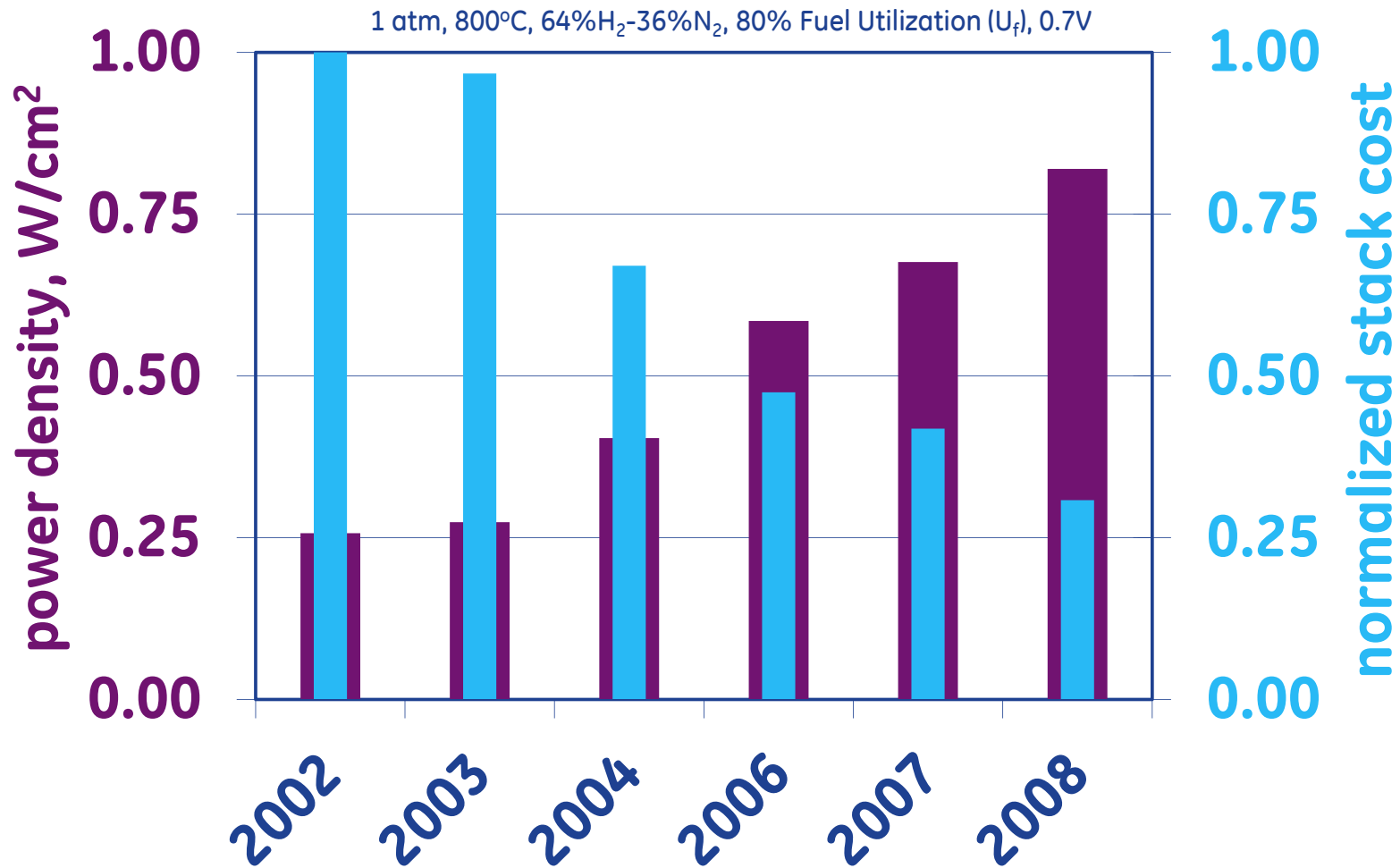
# Materials Set

# Anode Supported Solid Oxide Fuel Cell

	Layer	Function	Material	Thickness
Air Side	Interconnect	Gas & Electron Transport	Ferritic Stainless Steel	500 $\mu\text{m}$
	Protective Coating	Prevent interconnect Cr from poisoning cathode	$(\text{Mn,Co})_3\text{O}_4$	10 $\mu\text{m}$
	Cathode Contact Paste	Electrically connect cell with air interconnect	$(\text{La,Sr})\text{CoO}_3$	100 $\mu\text{m}$
	Cathode	Air electrode	$(\text{La,Sr})(\text{Co,Fe})\text{O}_3$	40 $\mu\text{m}$
	Barrier Layer	Prevent cathode Sr from reacting with electrolyte Zr	GDC $(\text{Ce}_{0.8}\text{Gd}_{0.2})\text{O}_2$	10 $\mu\text{m}$
Fuel Side	Electrolyte	Permit $\text{O}^{2-}$ transport, prevent air/fuel mixing	YSZ $(\text{ZrO}_2 + 8 \text{ mol } \text{Y}_2\text{O}_3)$	10 $\mu\text{m}$
	Functional Anode	Fuel electrode	NiO/YSZ	20 $\mu\text{m}$
	Anode Support	Mechanically supports Anode & Electrolyte	NiO/YSZ	200 $\mu\text{m}$
	Anode Contact Paste	Electrically connect cell with fuel interconnect	NiO	100 $\mu\text{m}$
	Interconnect	Gas & Electron Transport	Ferritic Stainless Steel	500 $\mu\text{m}$



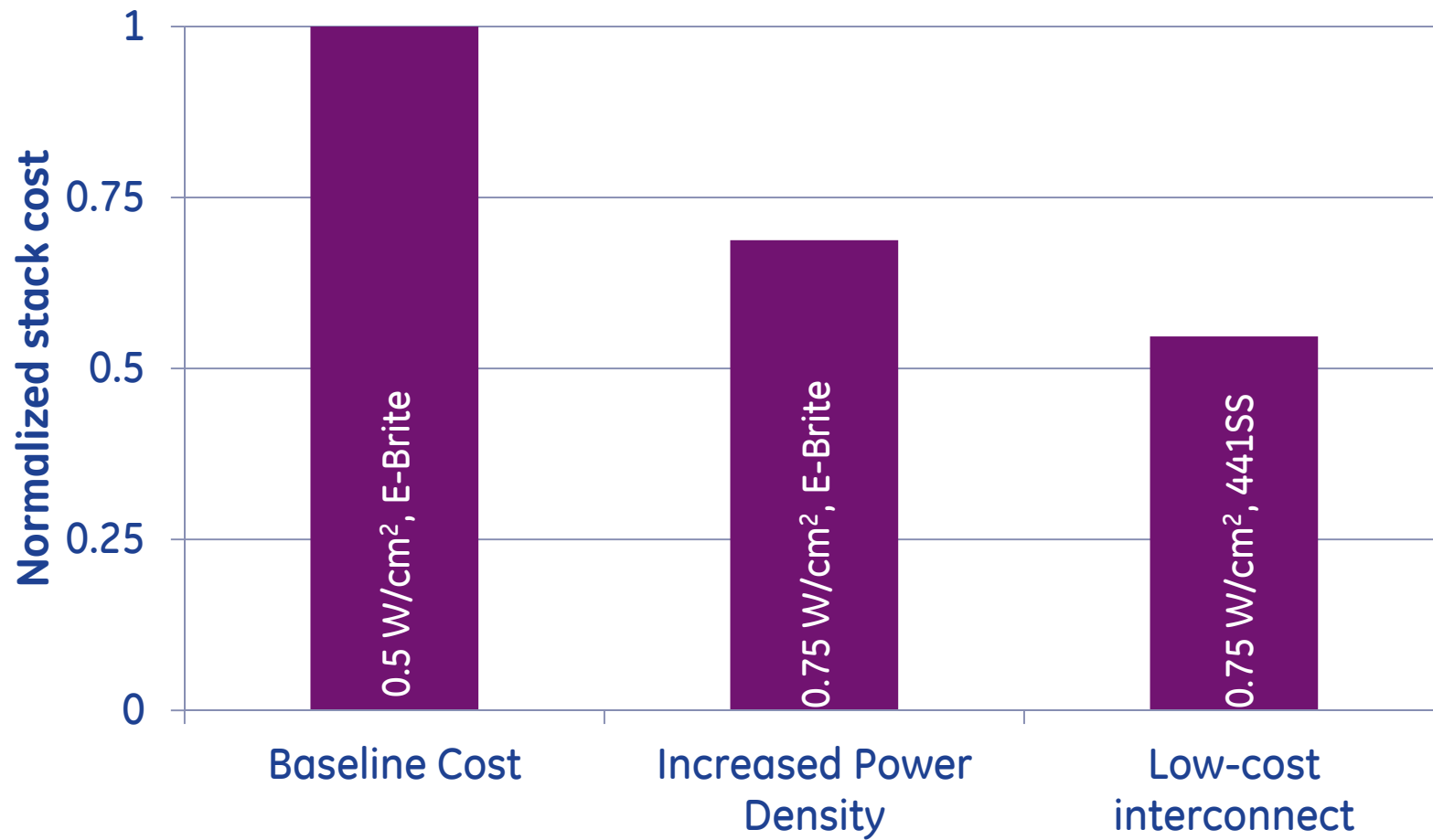
# Performance status



Significant performance improvement  
 Sufficient to meet cost targets, though higher  
 performance directly impacts stack cost

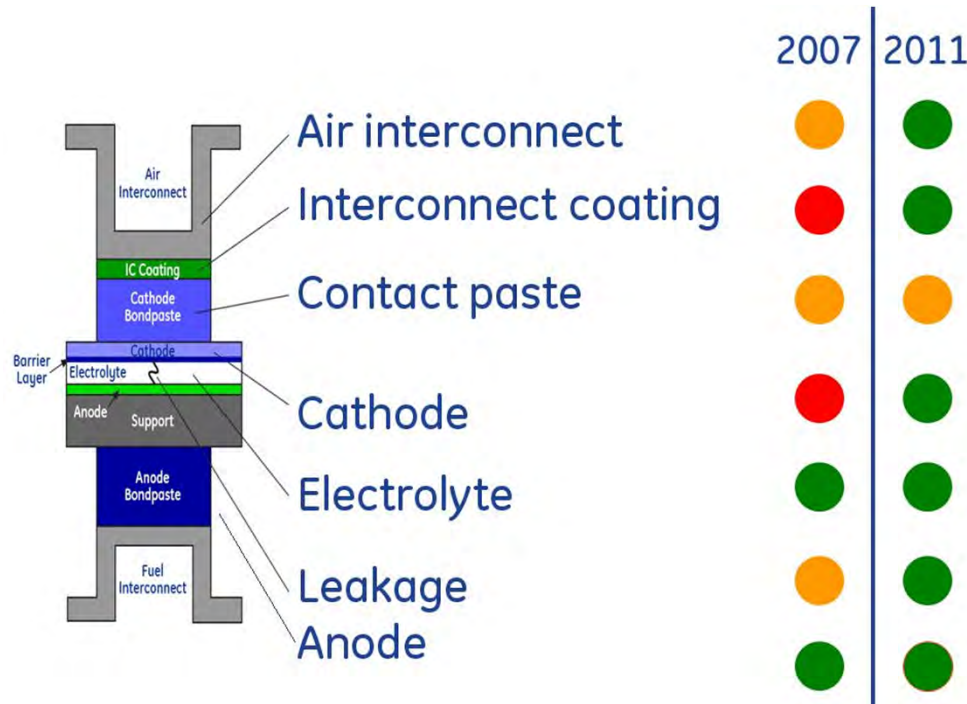


# Stack cost status

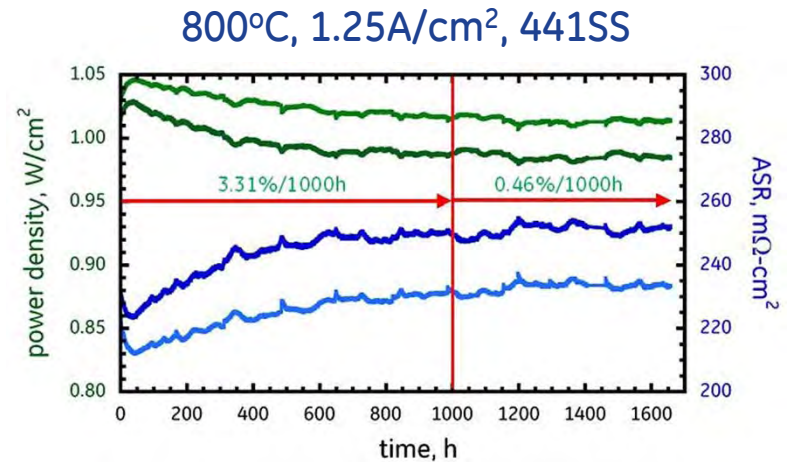


Cost reduction through performance improvement and interconnect alloy

# Degradation reduction



Degradation mechanisms identified and mitigation strategies validated

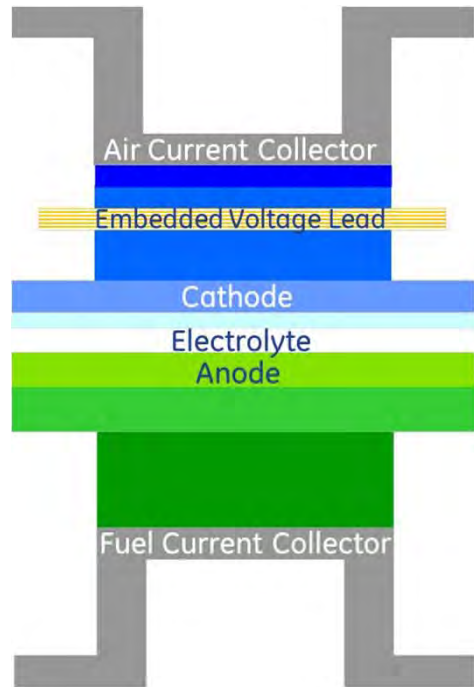


- Developed interconnect coating
- Stabilized cathode
- Validated low-cost interconnect alloy

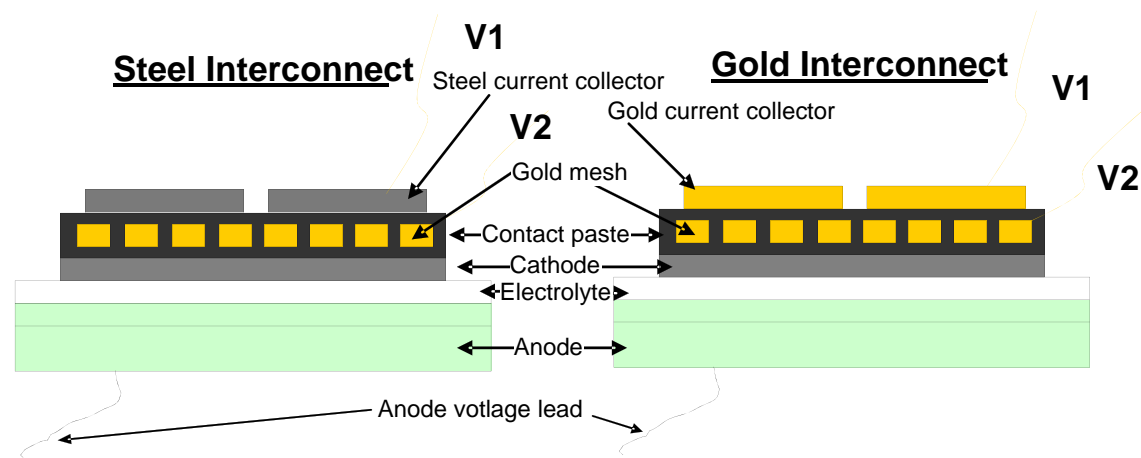
Cost and degradation risk significantly reduced

Engineering solutions exist, however, significant work remains to understand degradation fundamentals.

# Measurement of location-specific degradation



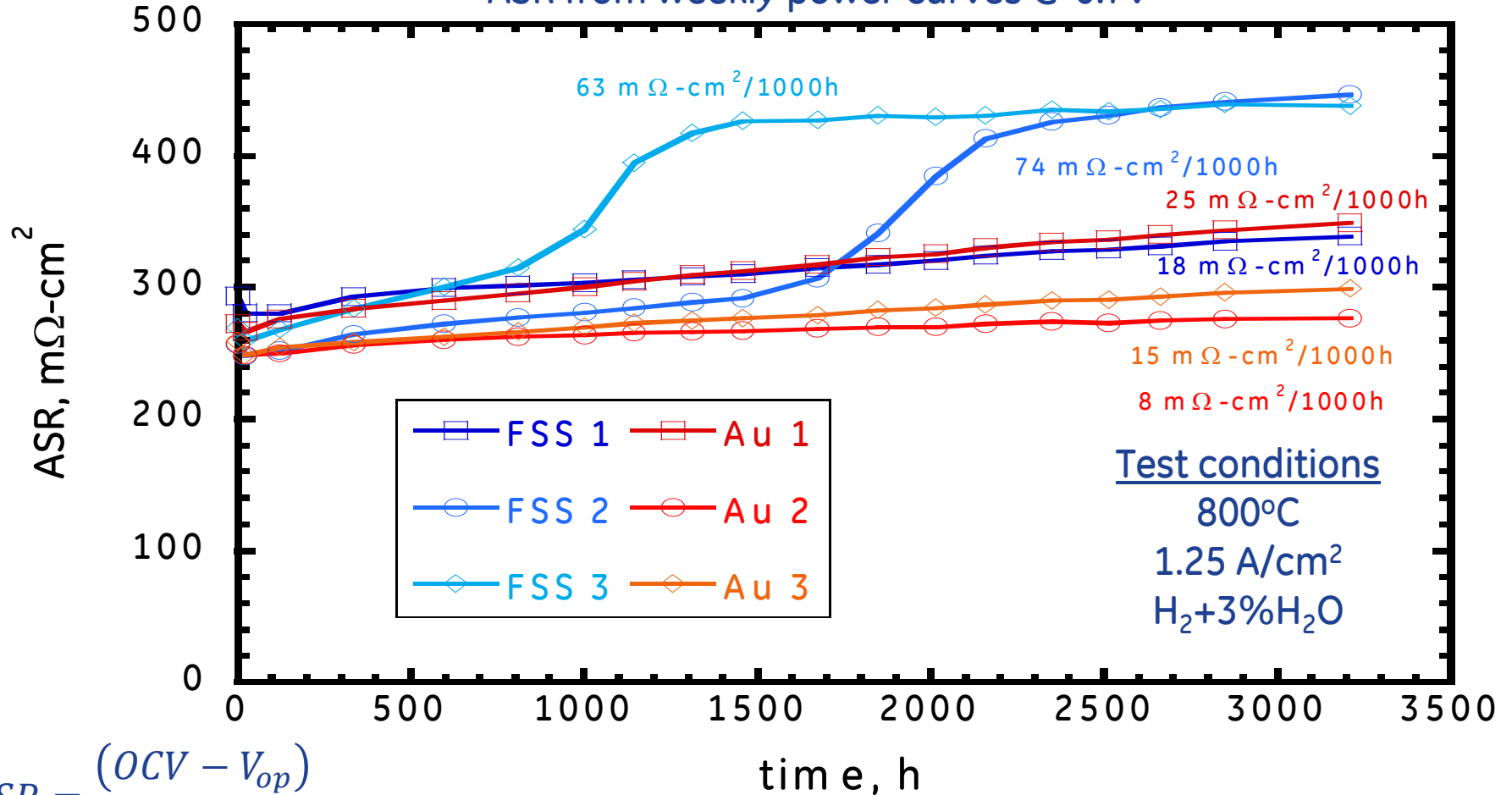
Test Conditions: 800°C  
1.25 A/cm<sup>2</sup>  
H<sub>2</sub>+3%H<sub>2</sub>O



Location-specific degradation can be measured from embedded gold mesh (voltage point)

# ASR degradation from power curves

ASR from weekly power curves @ 0.7V



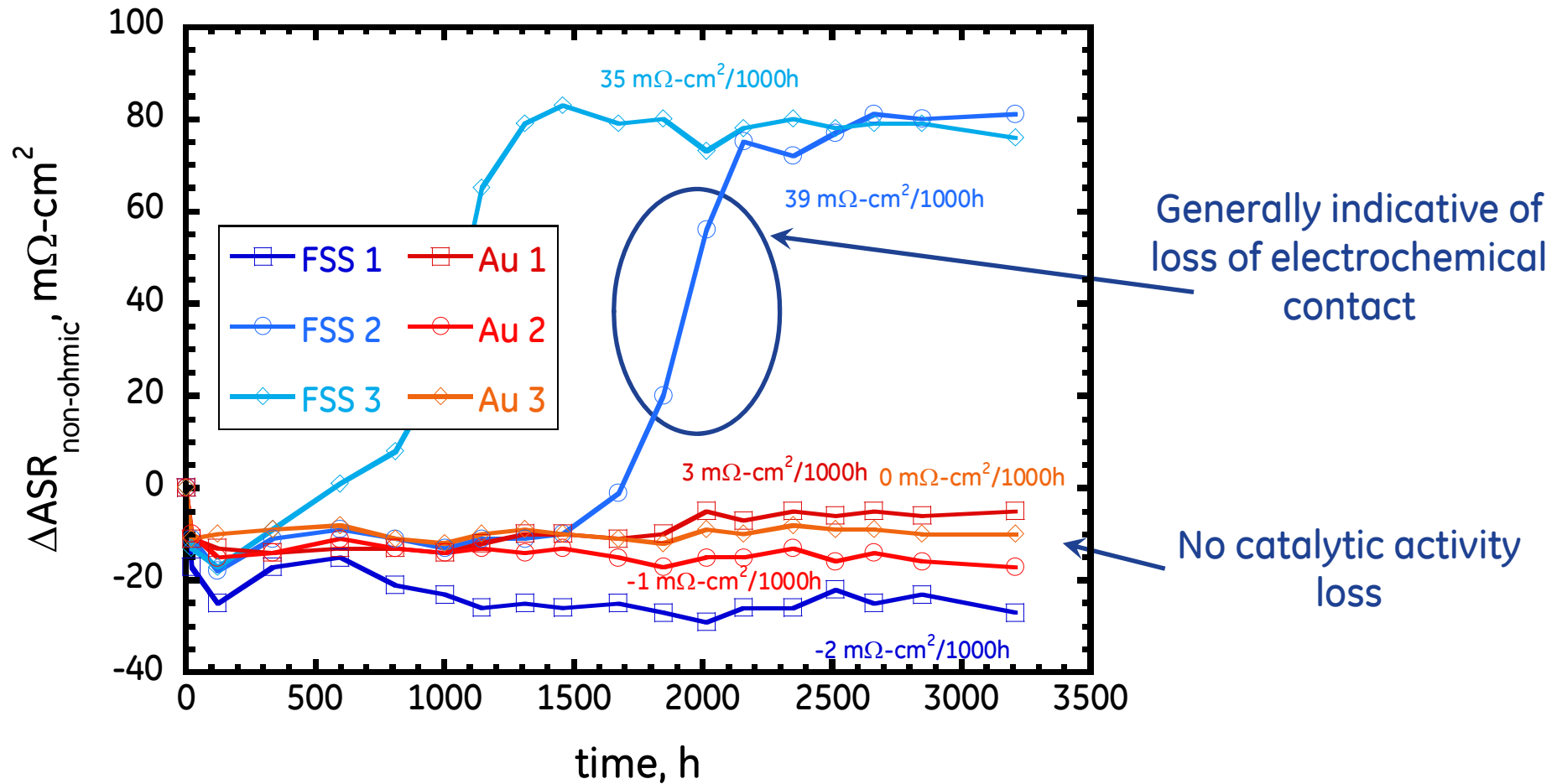
$$ASR = \frac{(OCV - V_{op})}{J_{op}}$$

ASR degradation consistent with traditional cell construction



# Non-ohmic degradation

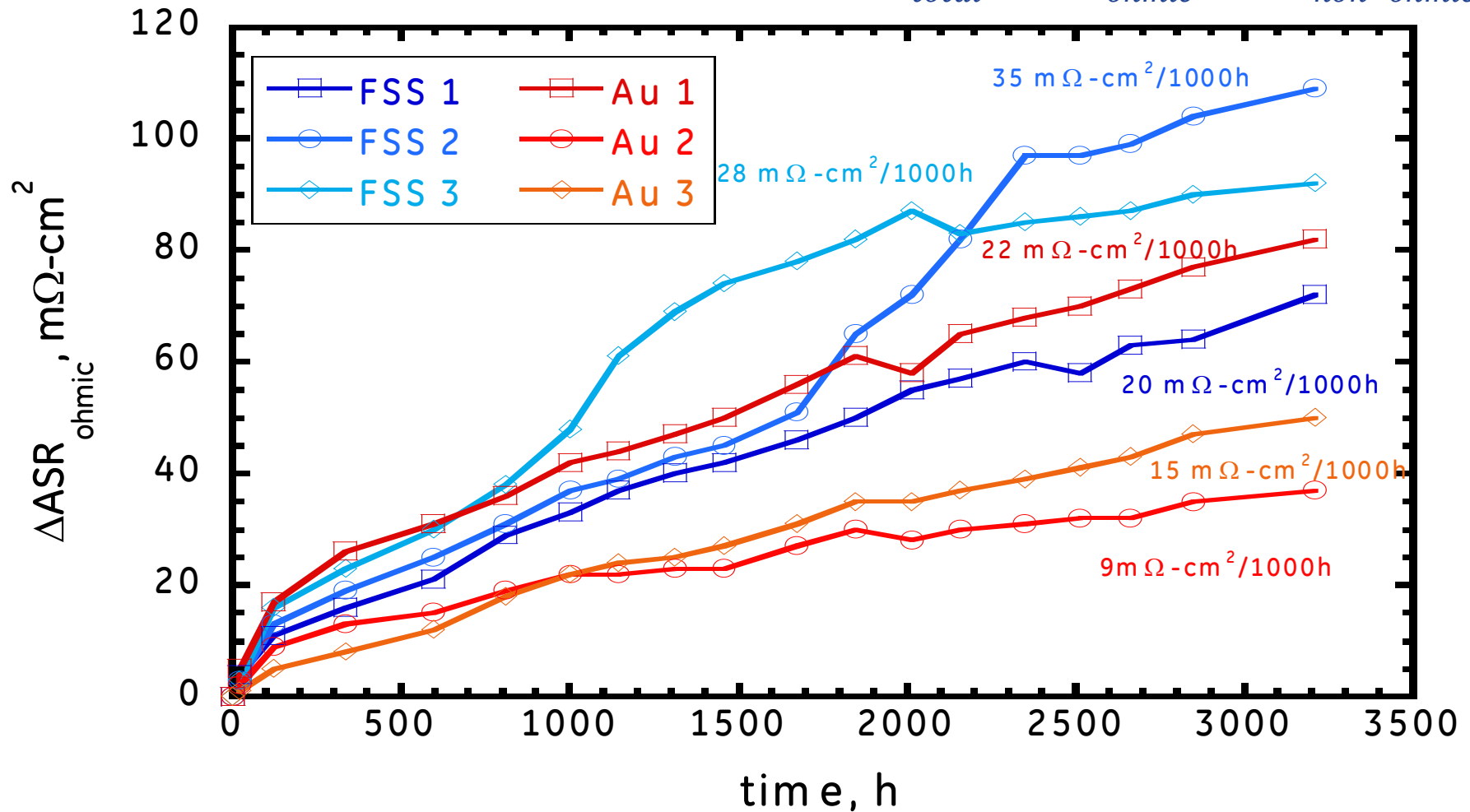
$$ASR_{total} = ASR_{ohmic} + ASR_{non-ohmic}$$



Non-ohmic degradation is nearly zero

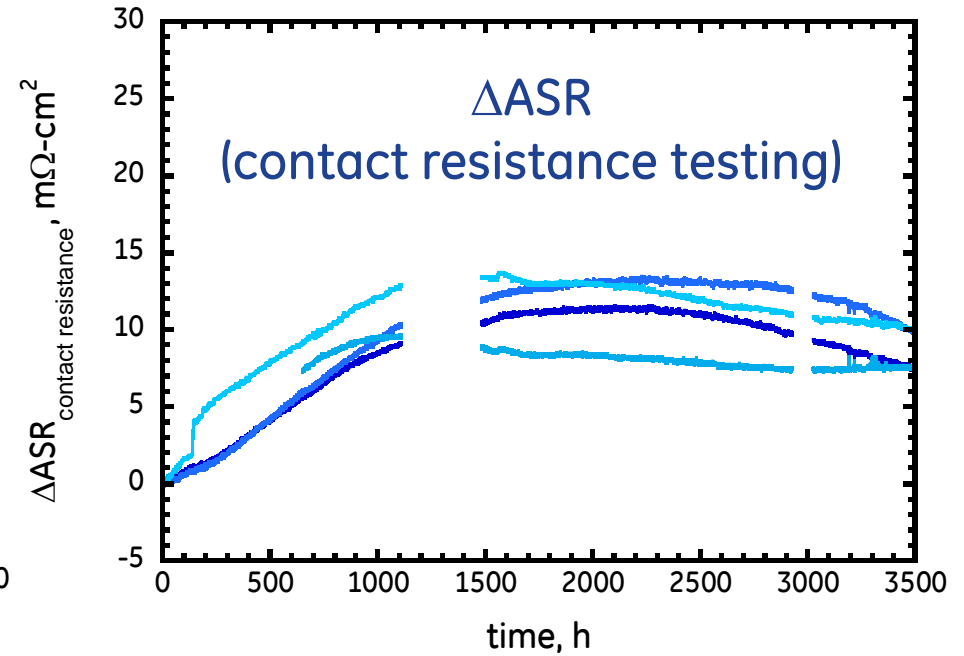
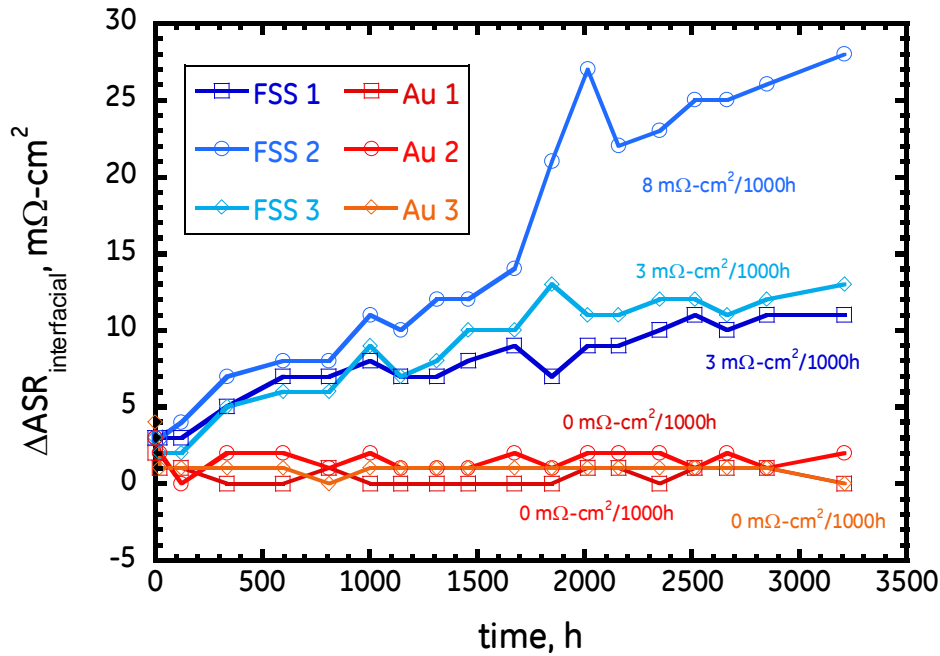
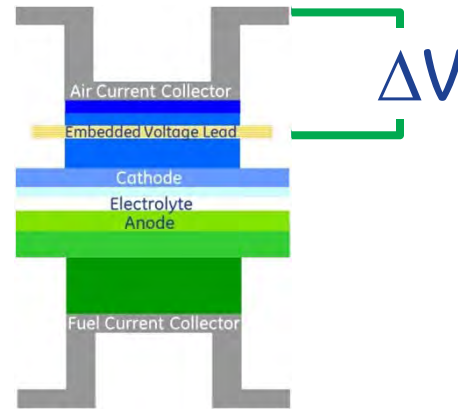
# Ohmic Degradation

$$ASR_{total} = ASR_{ohmic} + ASR_{non-ohmic}$$



Measured degradation is nearly all  
ohmic in nature

# Interfacial ASR

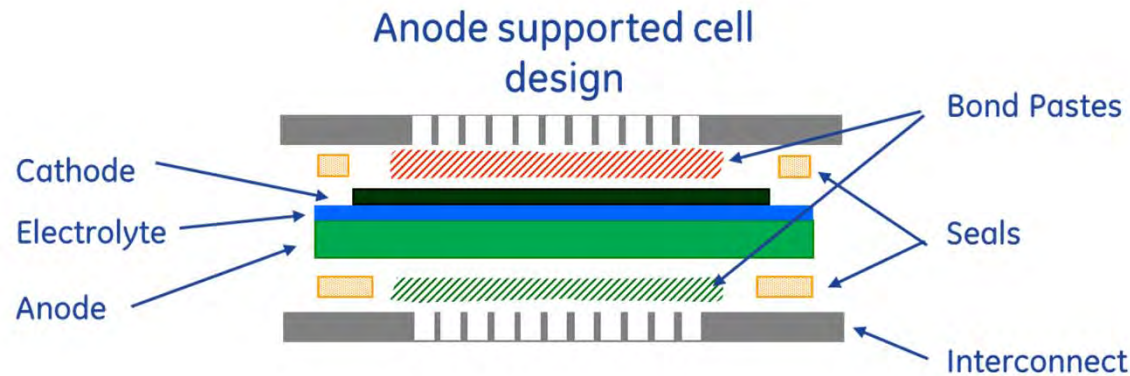


Confirms contact resistance data

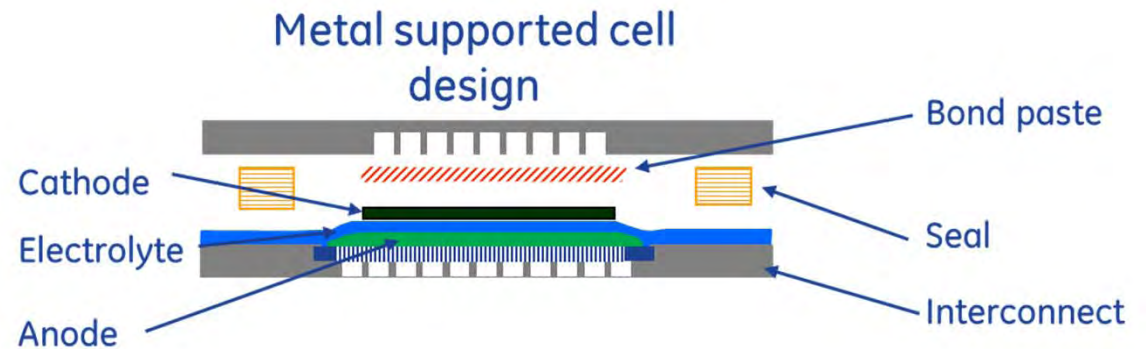


# Manufacturing

# Metal supported cell



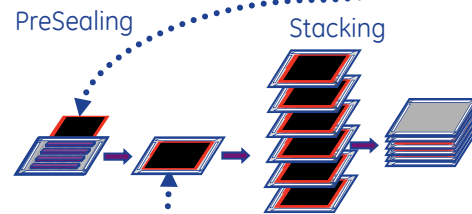
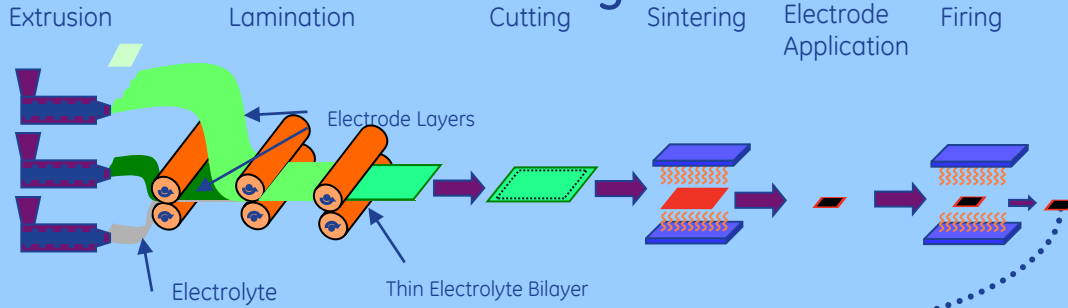
Advantages:  
 Integrated anode seal  
 Electrolyte in compression  
 Improved anode electrical contact  
 Increased active area  
 Lower anode polarization  
 Allows redesign of structures



Challenges:  
 Dense / hermetic electrolyte  
 Porous metal substrate degradation

# Low-cost manufacturing

## Sintered Cell Manufacturing



## Advantages

- Larger area / Scalable
- Simplified sealing
- Low Capex / Modular
- Lean Manufacturing

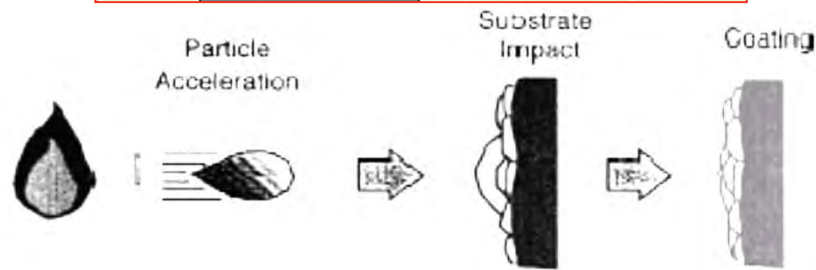
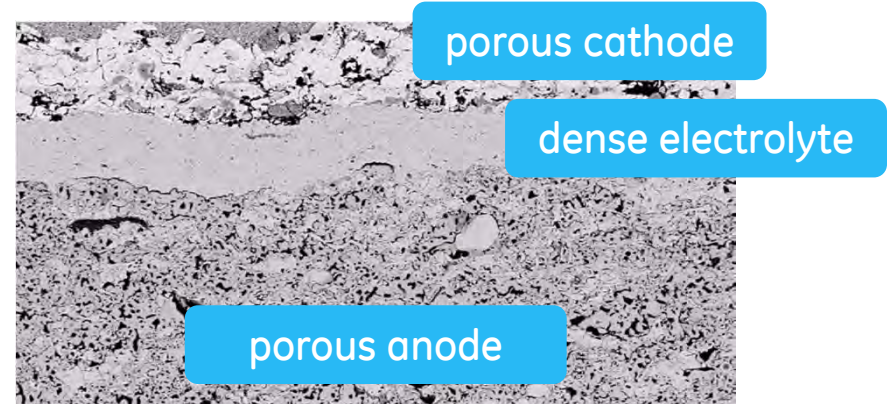
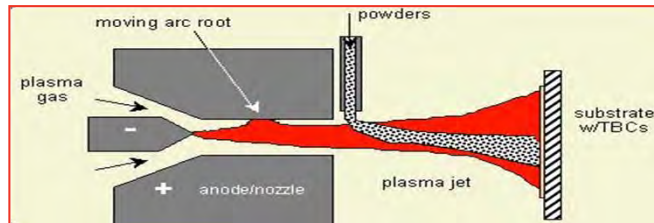
## Thermal Spray



Leverage GE thermal spray expertise

# Cell Manufacturing Processes

## Atmospheric Plasma Spray



Enable larger, thinner cells

Exploration of different processes and cell structures

Goal is to demonstrate scalability

Disruptive to stack manufacturing cost structure

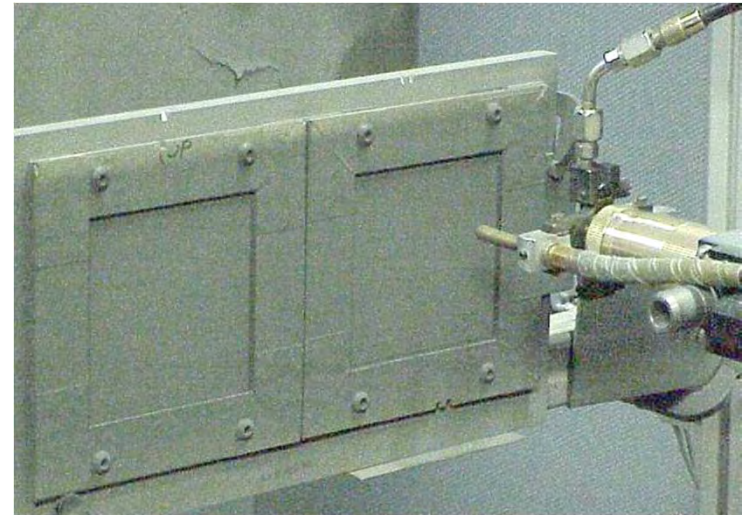
# Deposition Technology Progress

High throughput, many different structures / compositions / formats can be tested

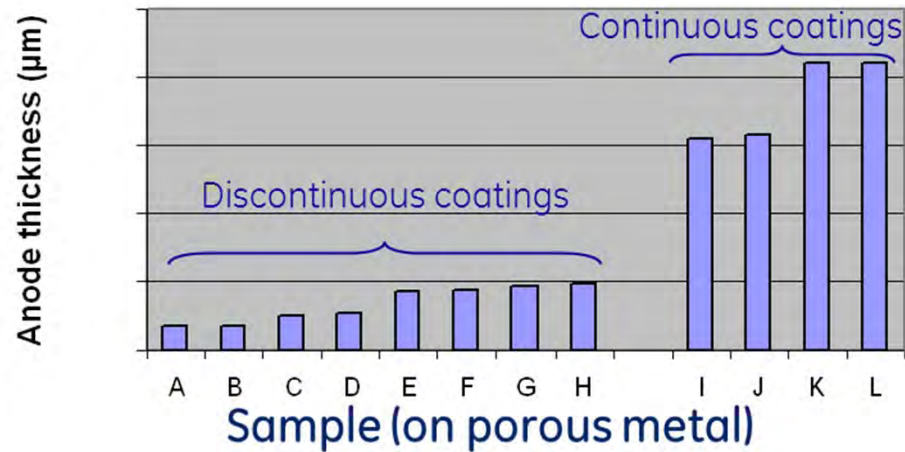
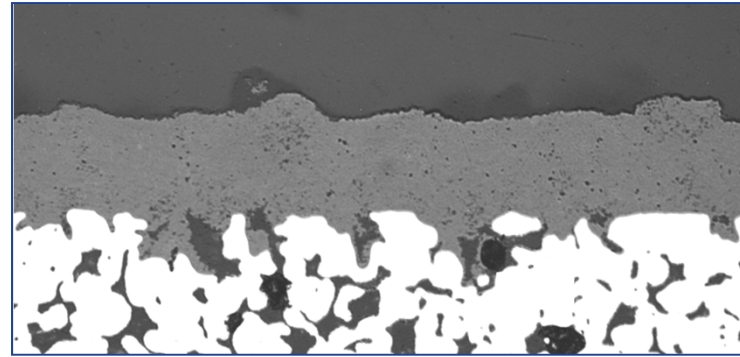
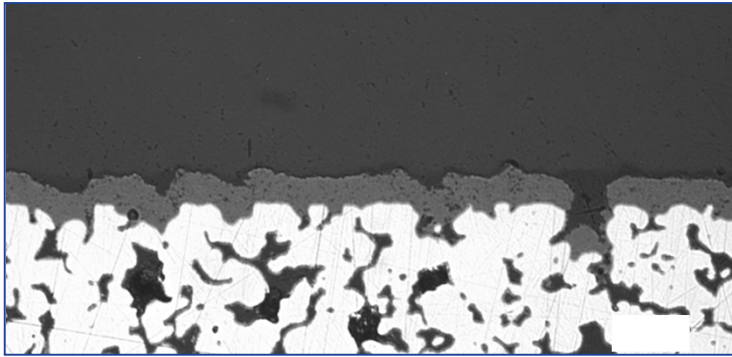
Cell and stack design tailored to deposition processes

Performance reaching sintered cell levels

Scale-up to 4" and 12" cell on-going

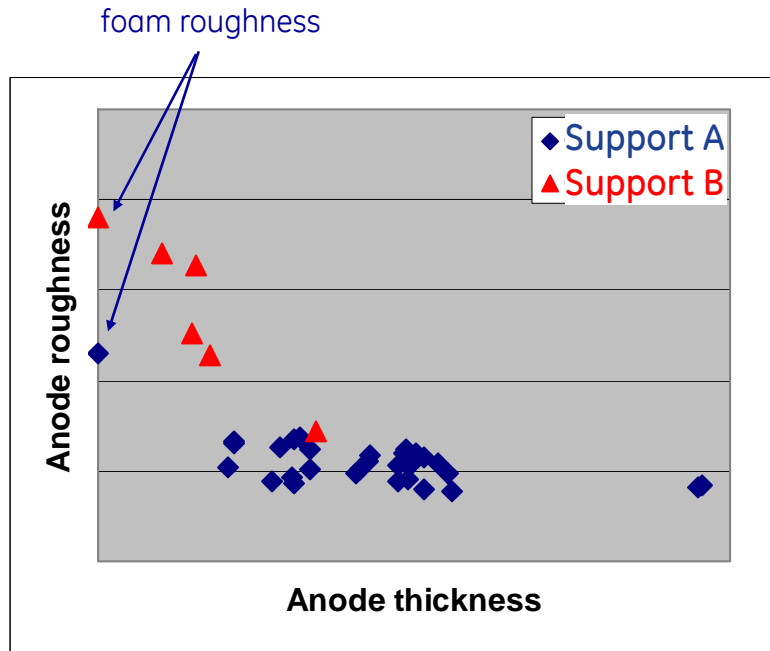


# Smooth Anode Development

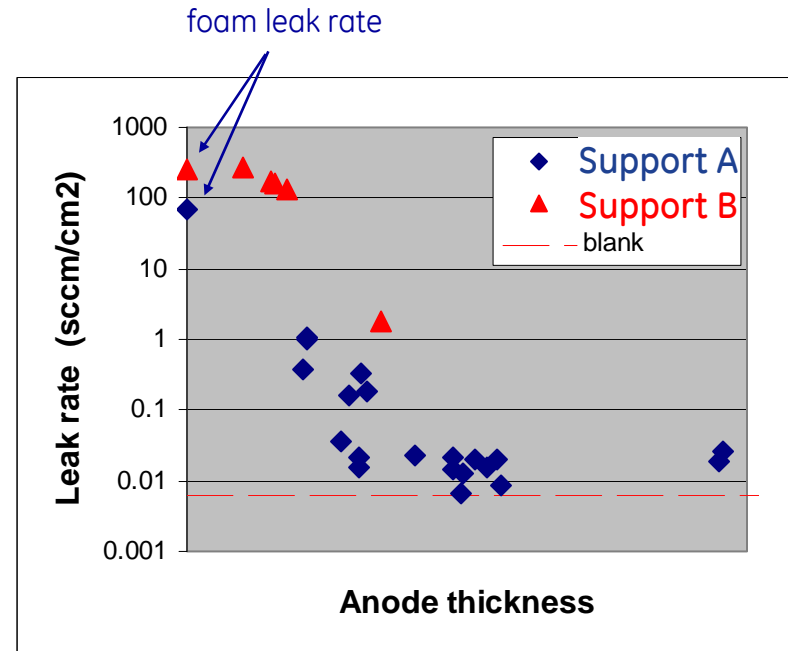


Minimum thickness required for deposition of a continuous anode on a porous substrate.

# Smooth Anode Development



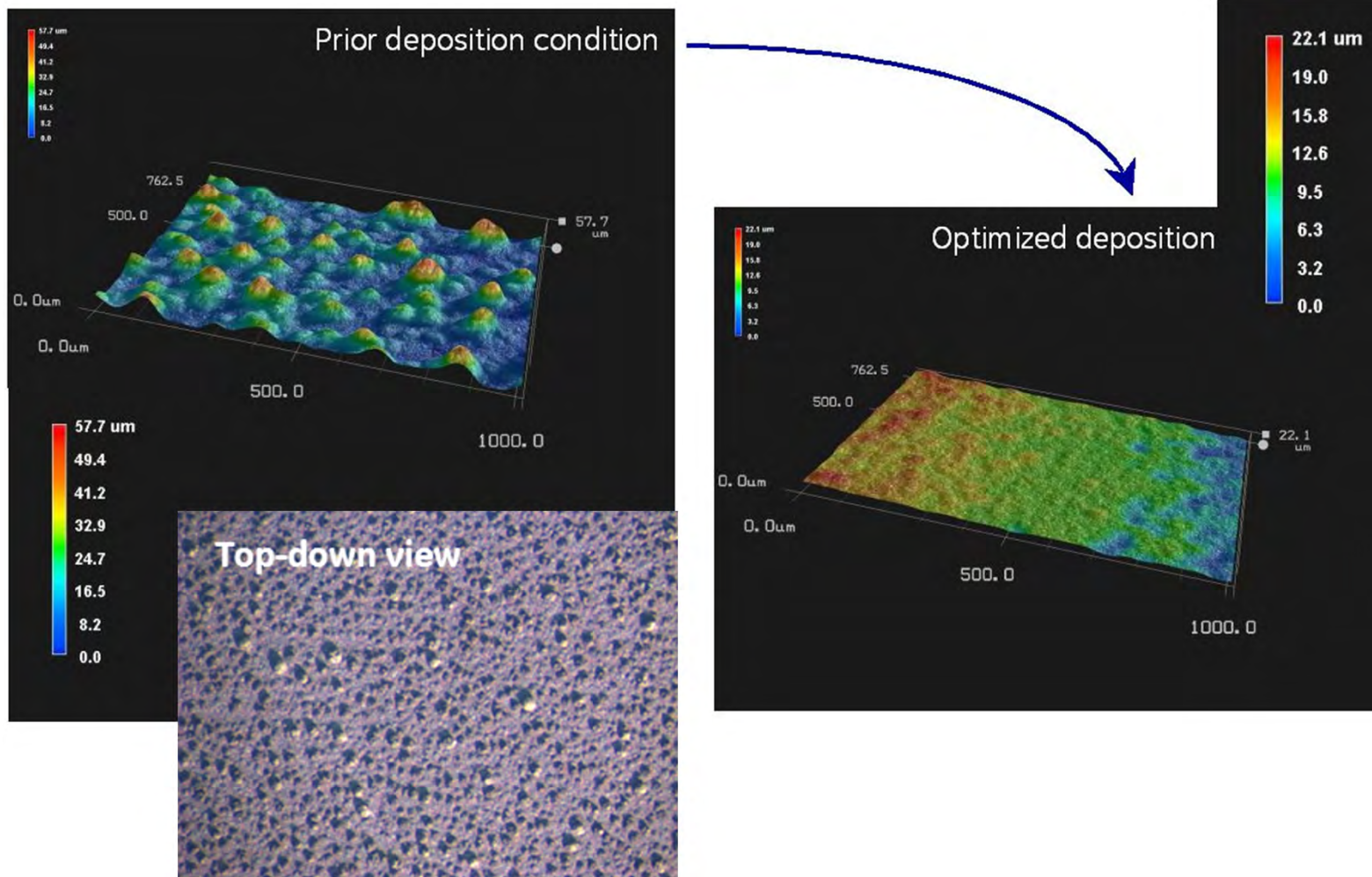
Anode roughness initially decreases with increasing thickness, then stabilizes to a fairly constant value. Transition is indicative of full foam coverage.



Leak rate decreases with increasing anode thickness until it stabilizes, which is indicative of full foam coverage.

## Minimum thickness established for full foam coverage with smooth anode

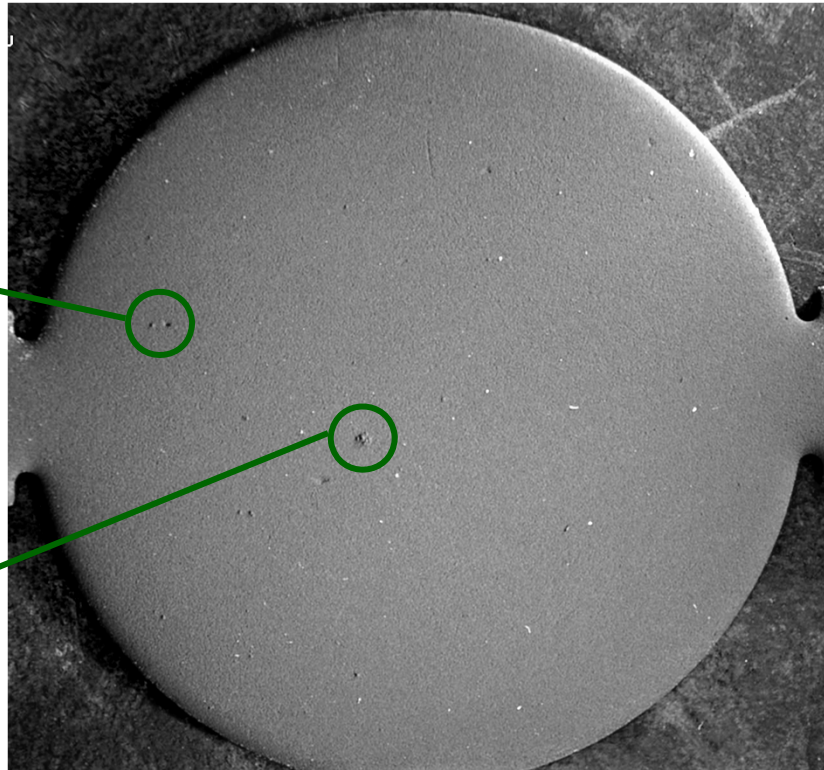
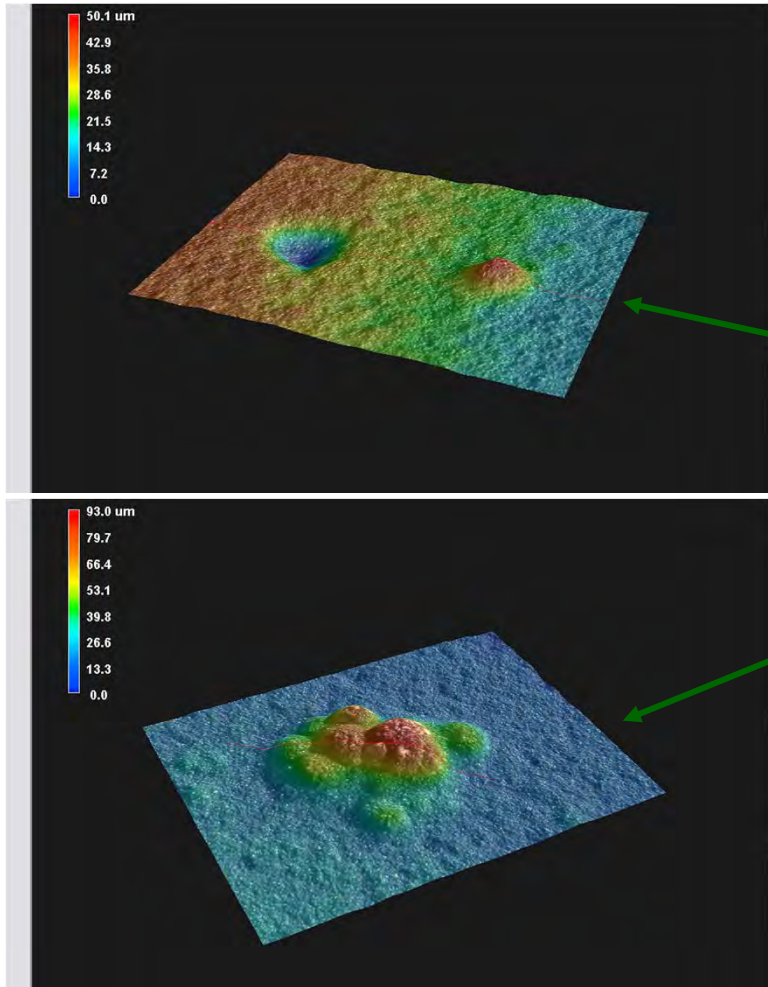
# Smooth Anode Development



Deposition condition developed for smooth anode deposition onto foam support



# Smooth Anode Development

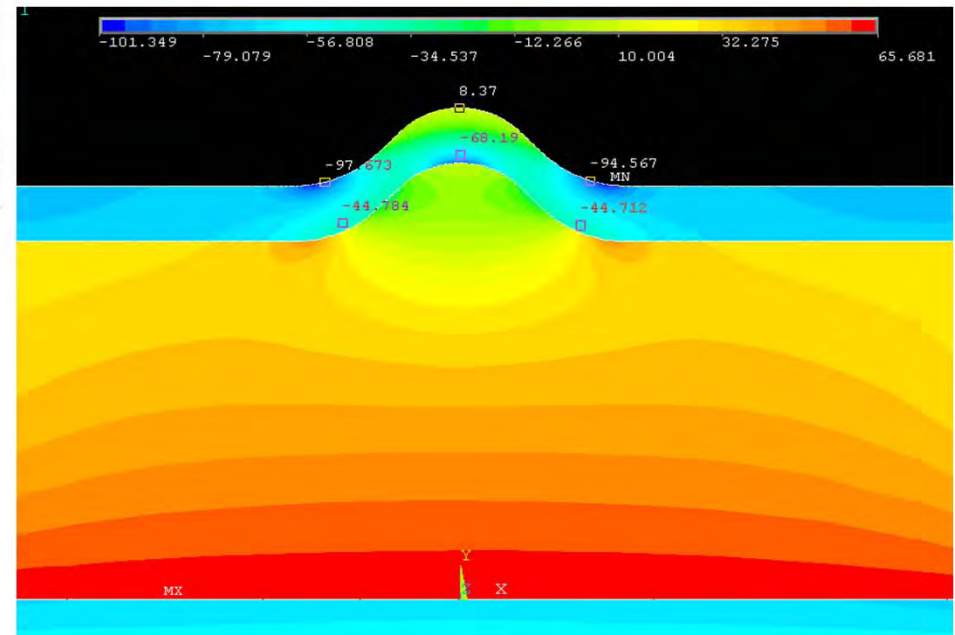
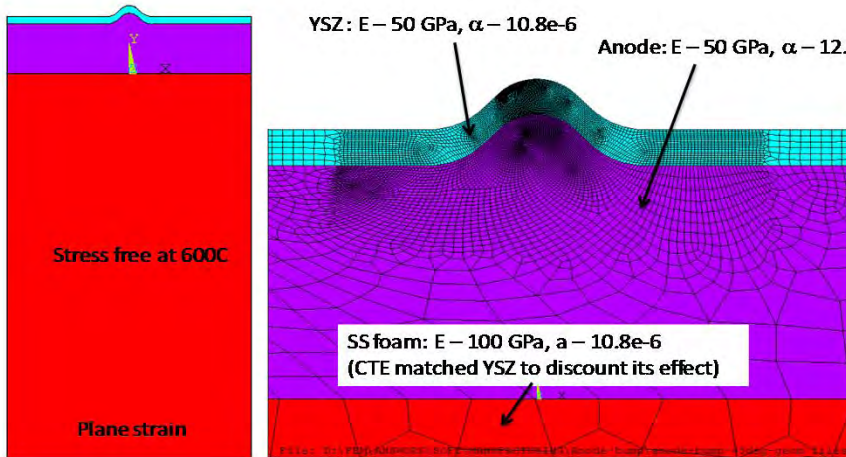


1" diameter foam coated with smooth anode. Surface is generally smooth, however a few localized defects can be identified.

Manufacturing defects may impact cell performance

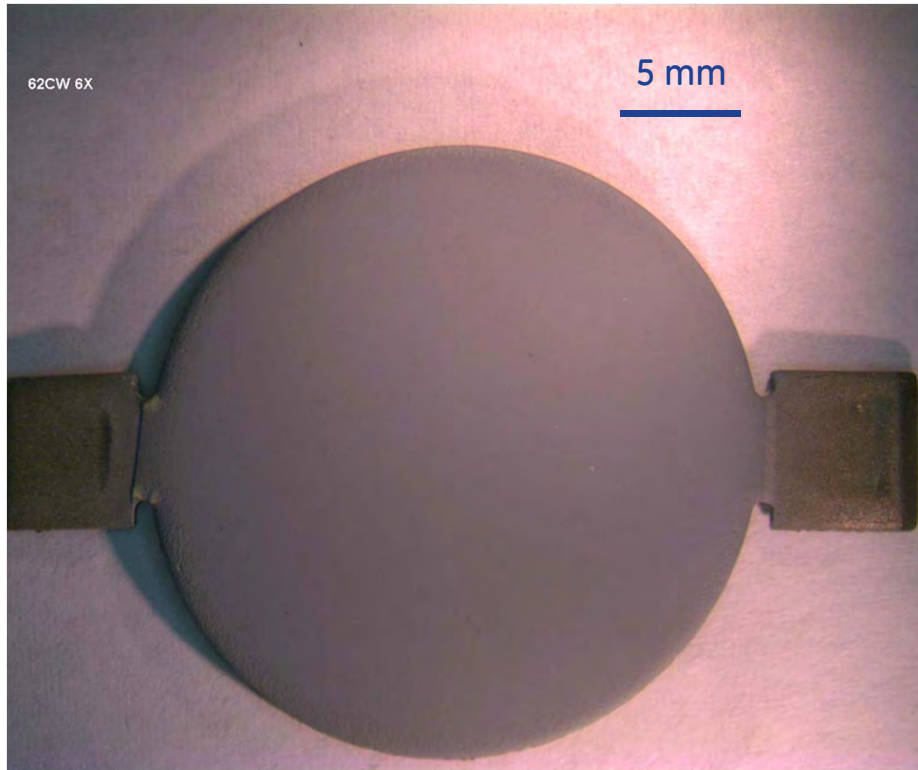
Improved process control for removal

# Effect of surface asperities on stress-state of thermal sprayed coating



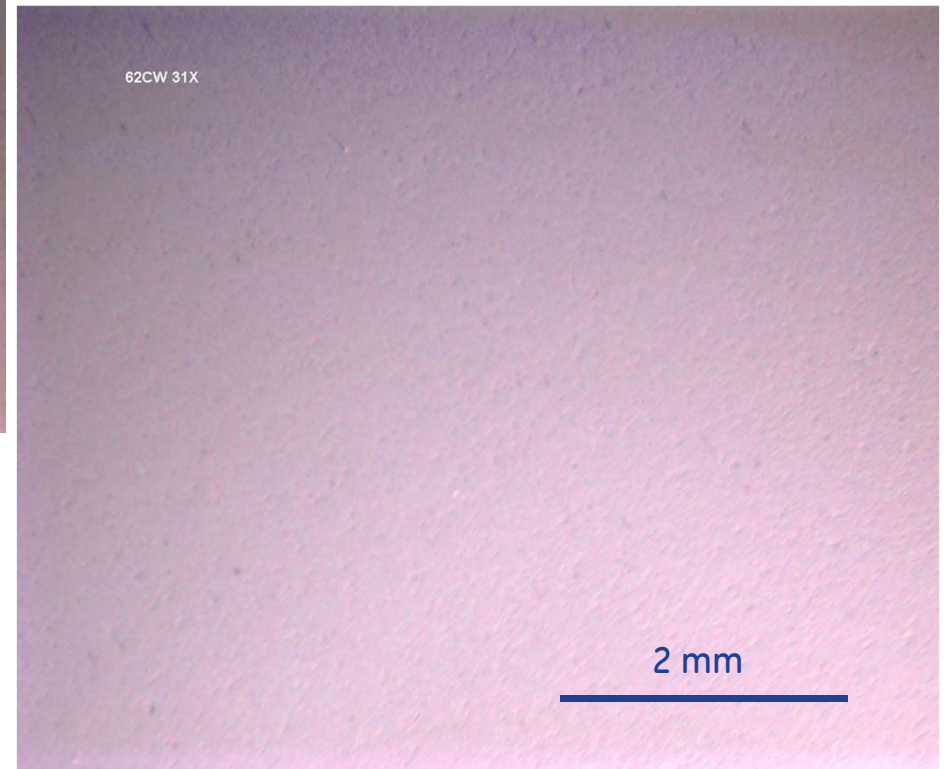
- Asperities cause local stress gradients in electrolyte
- Effect enhanced after anode reduction

# Electrolyte optimization

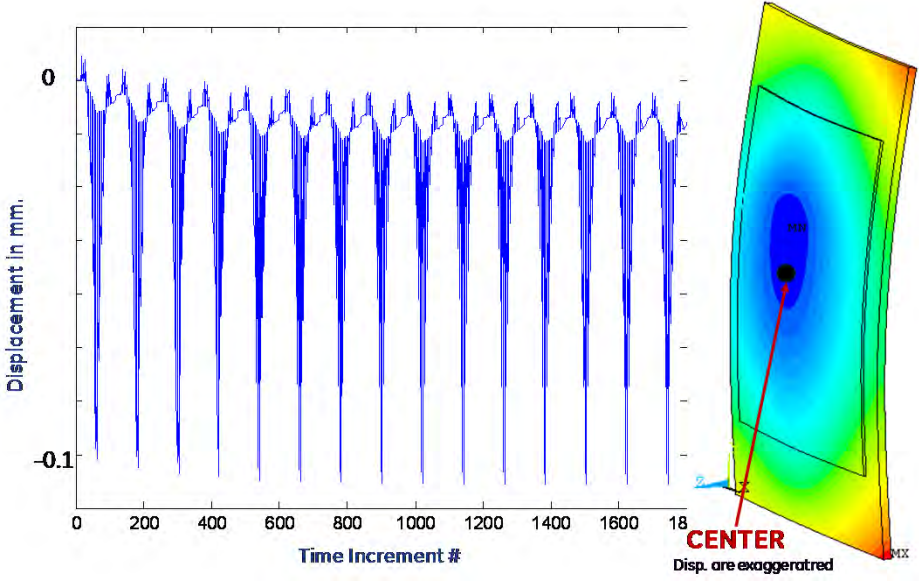
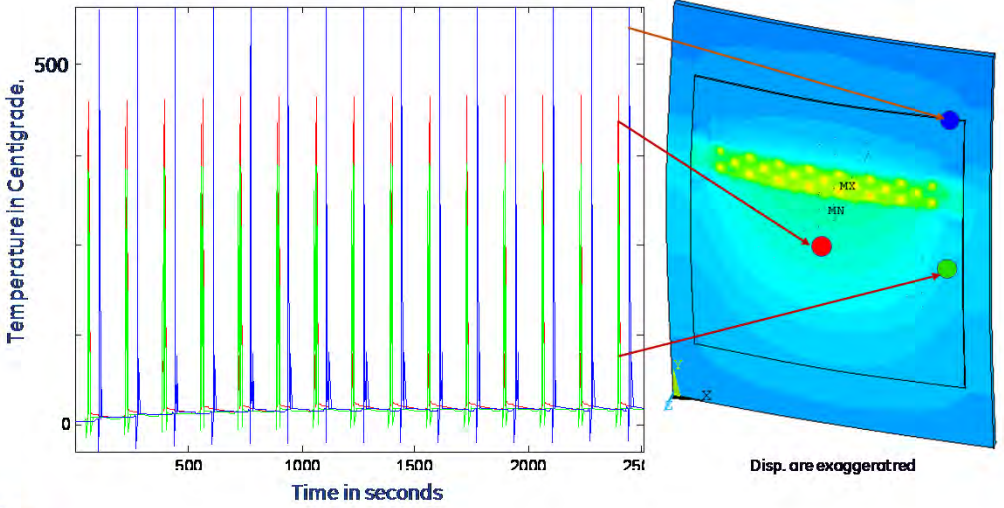
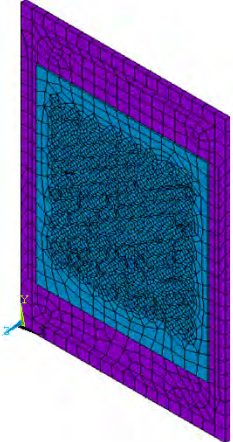


Top-down view of deposited coating on 1" diameter button cell

With a smooth anode & control of defects, high quality electrolytes can be deposited.



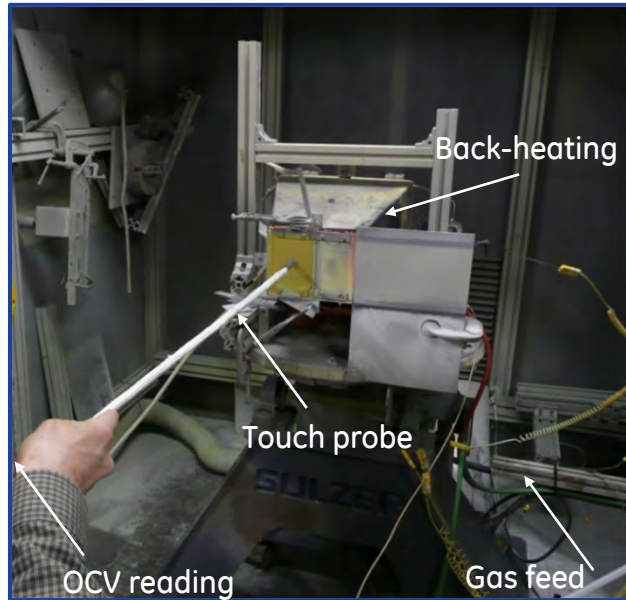
# Thermal spray heat flux modeling



Thermal heat flux modeling  
to support large area cell  
thermal management

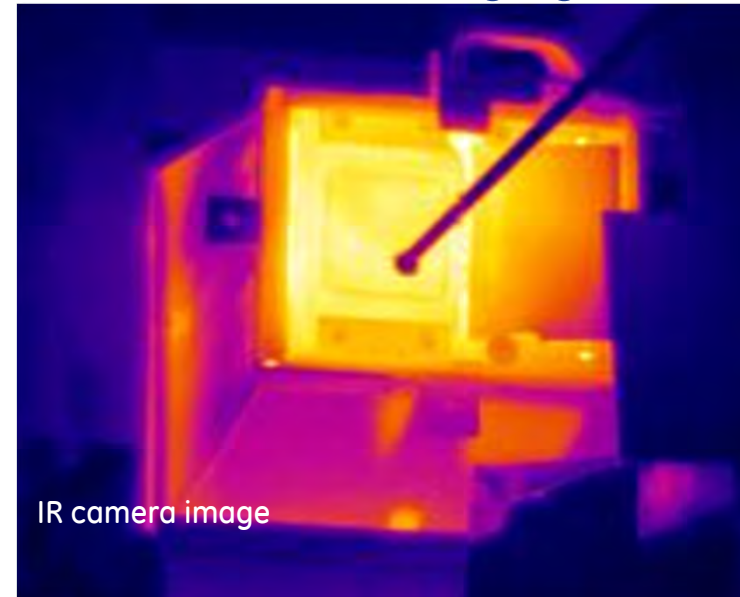
# In-situ characterization for thermal sprayed cells

OCV measurement



OCV mapping can be performed across entire cell

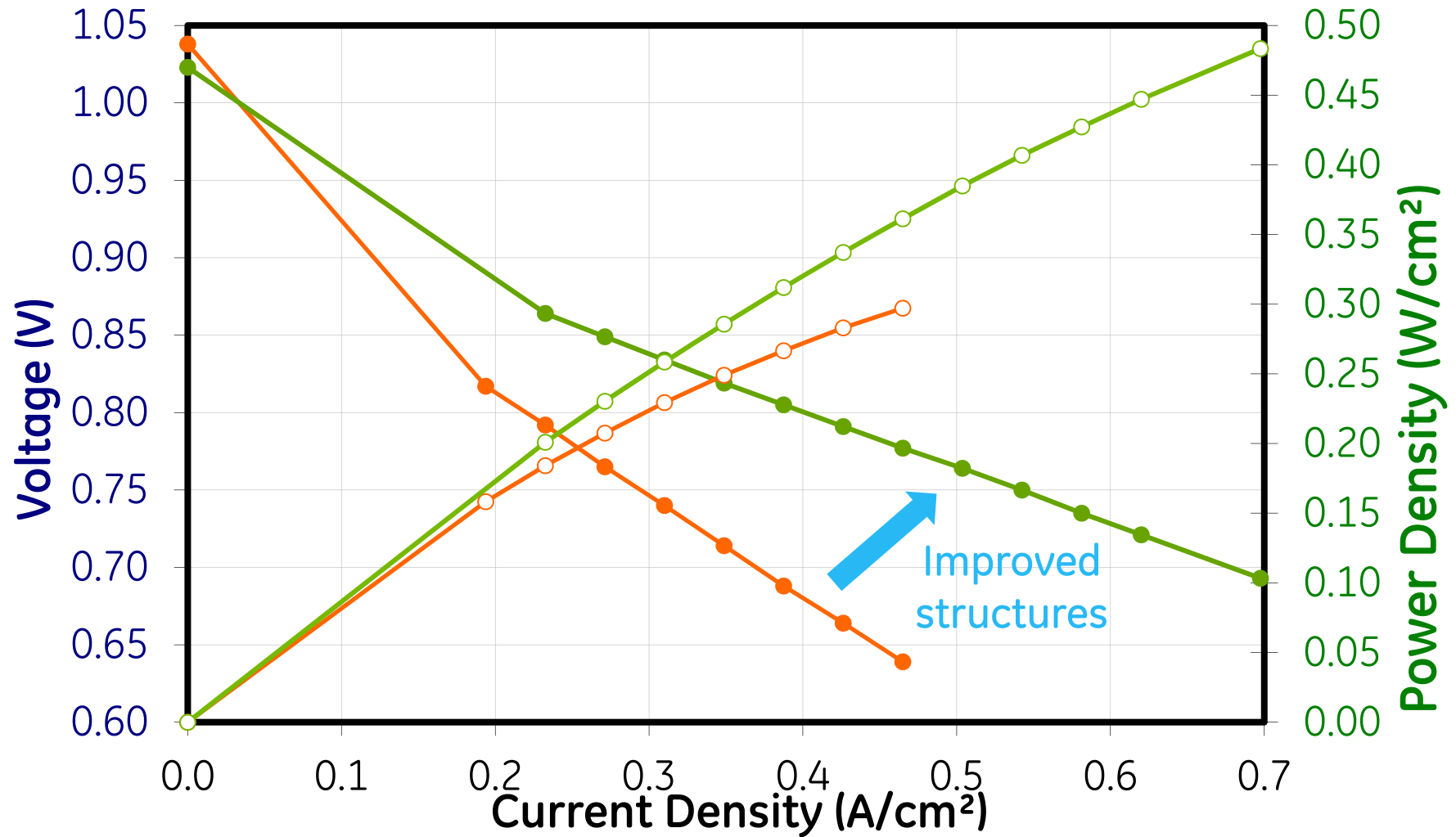
Thermal Imaging



Imaging during OCV testing helps in further cell characterization

In-situ measurements in the spray cell permit rapid feedback for development

# Thermal Sprayed Cells (25 cm<sup>2</sup>)



Optimizing electrodes and electrolyte for improved power density

# Summary

- Identified high-impact fundamental degradation mechanisms and developed cost-effective mitigation solutions.
- Demonstrated high, stable performance of LSCF-based cathode SOFCs with gold current collectors.
- Demonstrated parabolic power density degradation behavior with ferritic stainless steel (AL441HP) current collectors that is indicative of chromia scale growth.
- Implemented and tested an integrated thermal spray manufacturing system for SOFCs.
- Identified anode roughness as a key criteria to enable hermetic electrolytes.
- Developed thermal spray conditions to produce a smooth fuel electrode (anode) on a porous metal support.
- Demonstrated operational performance on 25cm<sup>2</sup> thermal sprayed cells.

# Acknowledgements

- Joe Stoffa, Briggs White, Travis Shultz, Heather Quedenfeld and Shailesh Vora of DOE/NETL
- Funding provided by the US Department of Energy through cooperative agreement DE-NT0004109.
- SECA partners
- GE SOFC Team

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